

Measurement of Diamagnetism in Water

Zijun Chen, E.D. Dahlberg, School of Physics and Astronomy

Introduction

- Diamagnetic materials create magnetic fields that opposes an externally applied magnetic field.
- Water is a diamagnetic material¹. However, the interaction is extremely weak, and visually noticeable effects can only be obtained using powerful superconducting magnets^{2,3}.
- In the past, demonstrations of diamagnetism in water using more accessible permanent magnets have been qualitative^{4,5,6}.

Goal

- The goal of the experiment was to quantify the diamagnetic effect of a permanent magnet on water using techniques accessible at a high school or introductory college physics level.

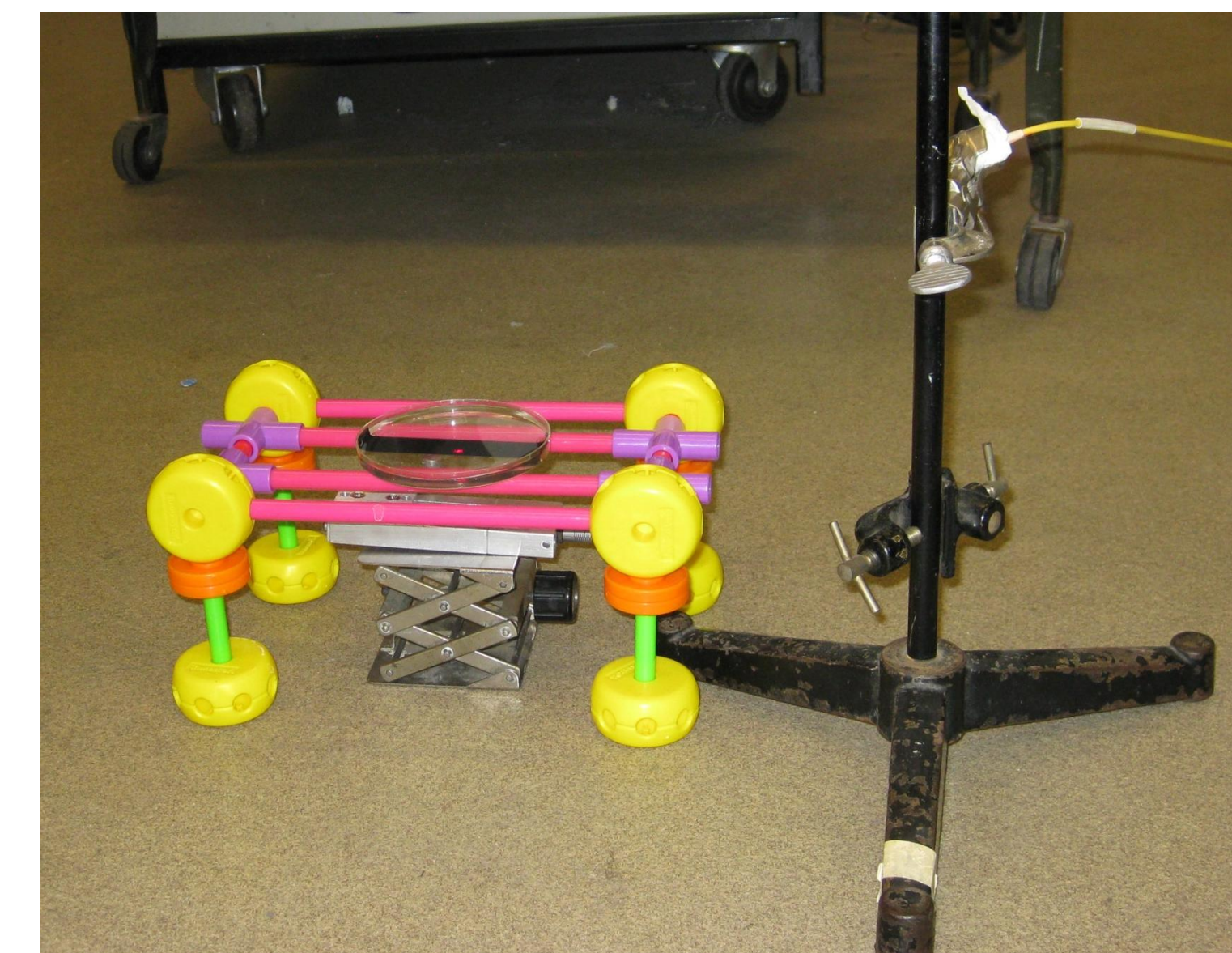
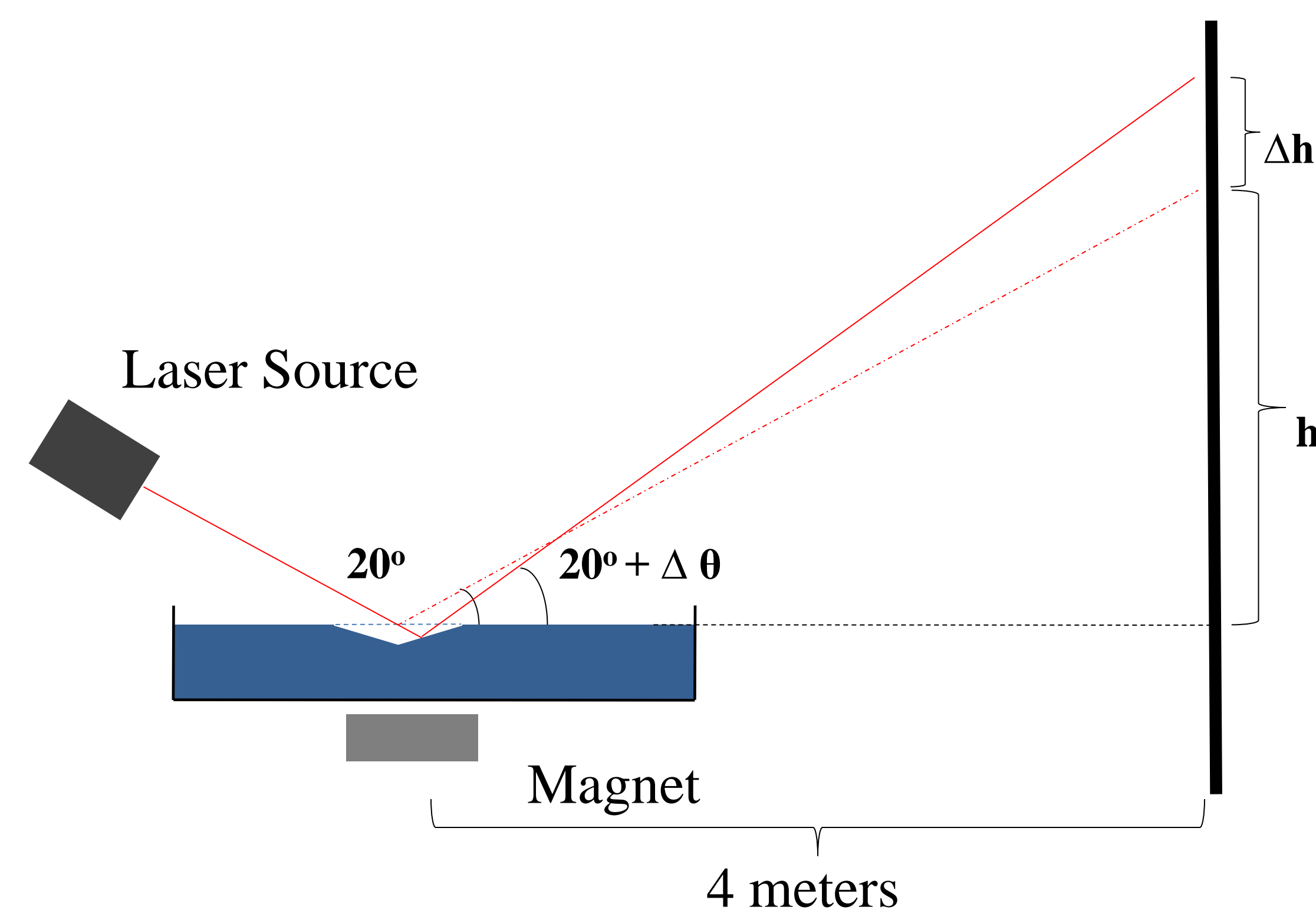
Theory

- Magnetic Energy Density²: $\mu = \frac{\chi B^2}{2\mu_0}$
- Gravitational Potential Energy Density: $\mu = \rho g \Delta h$
- By the law of conservation of energy, the two energies should be equal.

Methods

- A Petri dish was filled with water 0.25 cm deep. A neodymium (NdFeB) magnet was placed under the dish, creating a dimple in the water.
- A laser was reflected off the surface of the water, allowing us to measure the slope of the water. The position of the magnet was altered using a precision micrometer stage to obtain a profile of the dimple.
- The magnetic field profile of the magnet was measured using a gaussmeter. This allowed us to compare the energies of the magnetic interaction with the change in gravitational energy of the water.

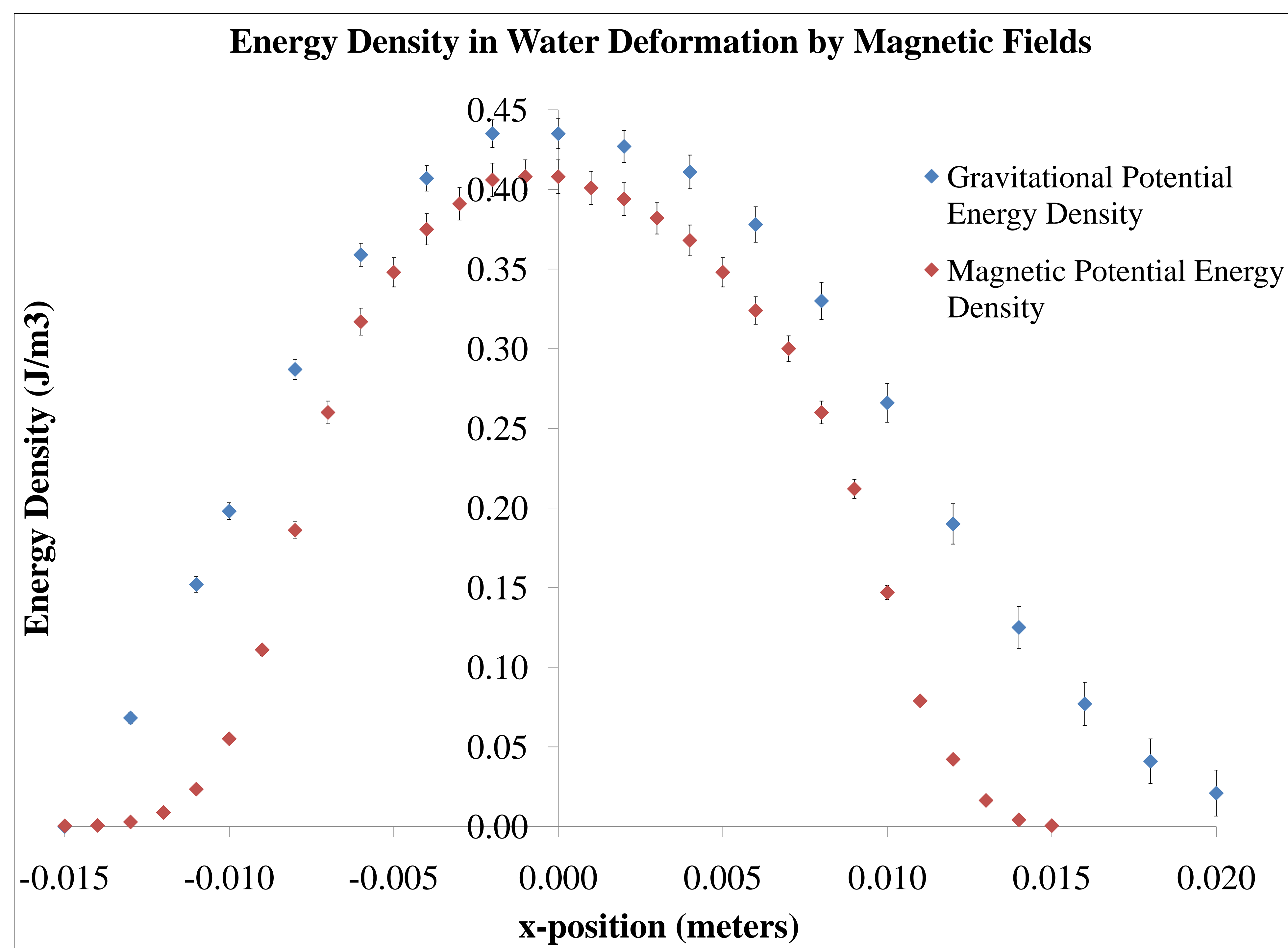
Apparatus



Conclusions

- At the maximum energy densities, the difference between the two energies is within two standard deviations.
- The magnetic field strength varies more than 5% for every 0.1 mm of variance in the measurement height, so the uncertainty on the magnetic measurements is likely underestimated.
- A better method for measuring the magnetic fields is required for an accurate comparison between the energies.
- The experiment shows that this method for measuring the diamagnetic effect is simple to carry out, precise, and can be accurate given some additional refinement.

Results



References

1. E. Beaugnon, R. Tournier, "Levitation of organic materials," *Nature* **349**, 470 (7 Feb. 1991).
2. S. Ueno and M. Iwasaka, "Parting of water by magnetic fields," *IEEE Trans. Mag.*, **30**, 4698-4700, Nov 1994
3. Noriyuki Hirota et al, "Rise and fall of surface level of water solutions under high magnetic field," *Jpn. J. Appl. Phys.* **34**, L991-L993 (1 Aug. 1995).
4. Charles A. Sawicki, "Small inexpensive diamagnetic levitation apparatus," *Phys. Teach.* **39**, 556-558 (Dec. 2001).
5. C. Concery, L. F. Goodrich and T. C. Stuaffer "More Diamagnetism Demonstrations," *Phys. Teach.* **41**, 74-75 (Feb. 2003).
6. Ron Edge, "A simple diamagnetic levitation experiment," *Phys. Teach.* **41**, 122 (Feb. 2003).

Acknowledgements

Thanks go to Prof. Dahlberg and the Undergraduate Research Scholarship program.