

Mass Production of Algae to Make Biodiesel

Steven A. Biorn

Faculty Mentor (Dr. Roger Ruan)

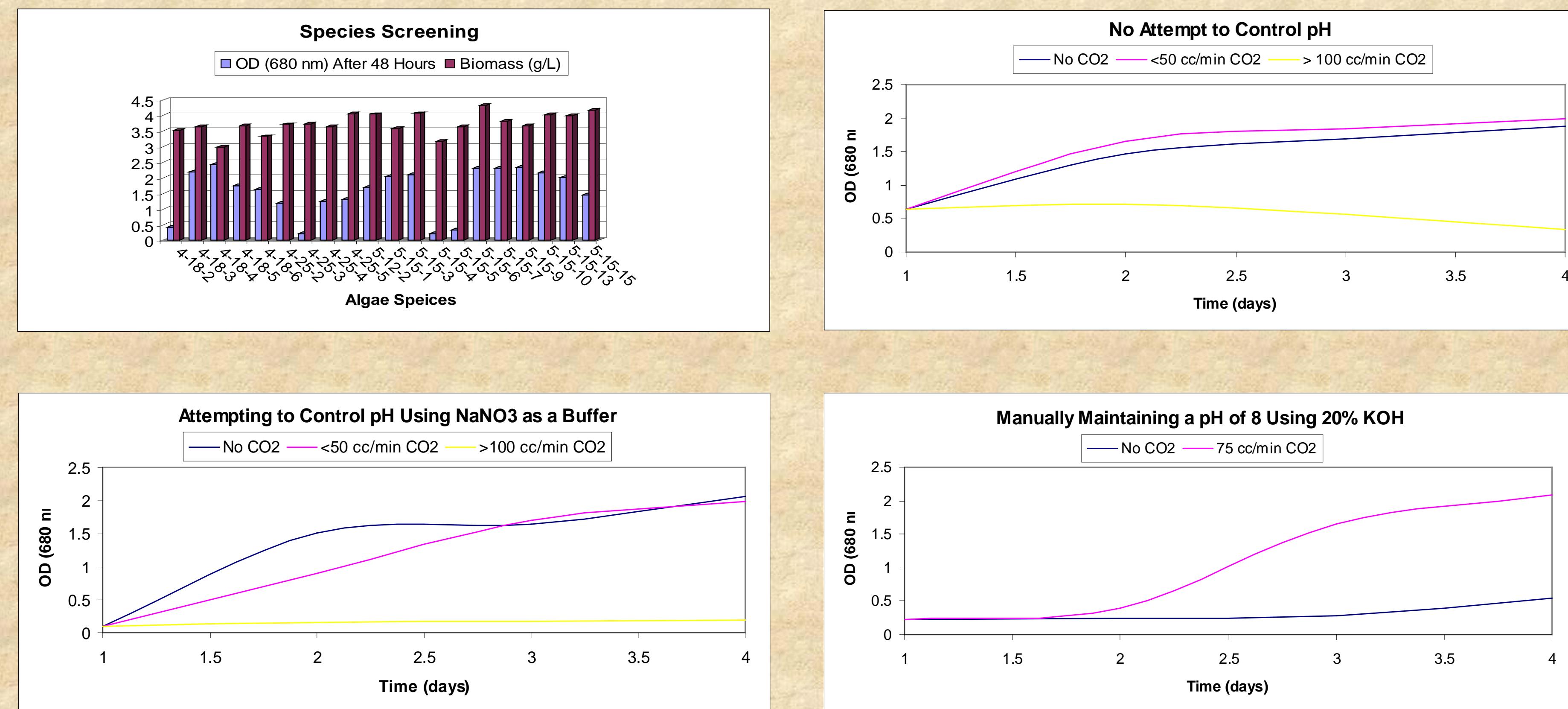
Bioproducts and Biosystems Engineering

University of Minnesota - Twin Cities

Objective: The purpose of the project is to isolate a species of algae with large quantities of triacylglycerides, starch and biomass. The Ruan Laboratory at the University of Minnesota has isolated several species of algae from Minnesota's ponds, rivers and lakes. Variations in CO₂ quantity, and how it affects algae growth were also investigated.

Why would algae make a good biofuel alternative?

- Use up to 10 percent of solar energy to make cellular structure (1)
- 45 to 220 times the triacylglyceride production of terrestrial biomass (3)
- 7 to 60 percent dry weight is triacylglycerides (2)
- Some species of algae can double their biomass every 24 hours (4)
- Feasible to grow enough to meet U.S. fuel demand, using only 3-6 percent of total crop area (4)
- Also produces methane from anaerobic digestion of biomass after triacylglyceride extraction (5)



Discussion and Future Work:

From what is known about the reaction of photosynthesis; $6\text{CO}_2 + 6\text{H}_2\text{O}$ (+ light energy) $\text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$, it was expected that supplying algae with more CO₂ would increase its replication rates. This prediction was made based on La Chateliers Principle, where an addition of reactants in the equation will drive the reaction to the right, therefore forming more glucose ($\text{C}_6\text{H}_{12}\text{O}_6$) and O₂. The experiments discussed above support this hypothesis, and also suggest that growth rates are heavily dependent on pH levels. Systems could be developed where pH is measured and maintained using a computer, or even simpler yet, base could be slowly dripped into the algae bioreactors at a given pace to counteract the decrease in pH caused by the addition of CO₂.

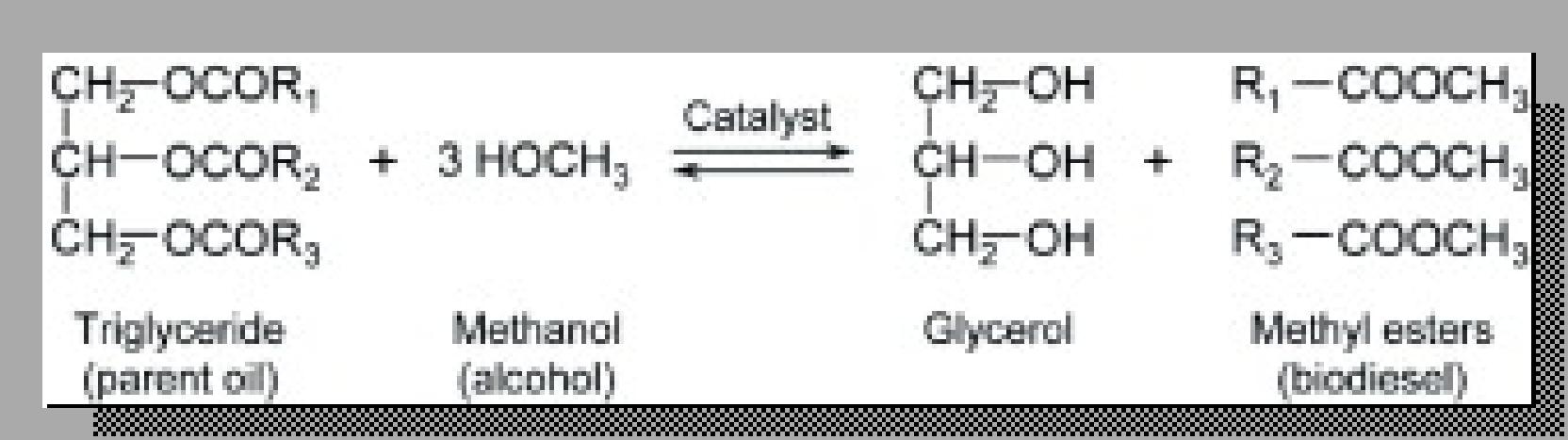
Acknowledgements:

A special thanks to the Initiative for Renewable Energy and the Environment (IREE), Center for Biorefining, and the Undergraduate Research Opportunity (UROP) program for providing funding for these experiments. Also thank you to Dr. Roger Ruan and Dr. David Kong for allowing me to work in their laboratory, offering insight and their years of experience to make this possible.

Methods and Results:

Twenty different species previously isolated by the Ruan laboratory were scaled up from solid media in a test tube to 100 ml of liquid media in 250 ml Erlenmeyer flasks. Light intensity, media concentrations, temperature and aeration times were kept constant for all species. The algae concentrations were measured every twenty four hours using a spectrophotometer (Spectronic Genesys-5, 680 nm-visible light). After eight days, biomass (g/L) was determined for all twenty species. The two species with the fastest growth (based on OD) and highest biomass were selected for the CO₂ augmentation study. Both CO₂ and atmospheric air were supplied to the algae at varying concentrations, as determined by a gas flow meter. The initial study found that while some CO₂ increased algae growth, too much CO₂ lead to the formation of carbonic acid and thus a decrease in pH levels. pH levels were measured using a Accumel AB15 pH meter. These low pH levels inhibited and in some cases were fatal to the algae. Past studies performed by the Ruan Laboratory at the University of Minnesota have shown a homeostatic pH range near 8. A second study was conducted using sodium nitrate (NaNO₃) as a buffer in an attempt to maintain the pH within the desired range. This study yielded similar results, where a little CO₂ and a lack of CO₂ showed similar growth rates, but too much CO₂ caused a large decrease in pH. From here, a third study was derived where pH would be taken each day and twenty percent potassium hydroxide (KOH) would be added to adjust the pH back to 8 for those flasks receiving CO₂. Manually maintaining the pH has shown promising results, where one species has OD measurements nearly four times that of the control, which did not receive any CO₂.

Biodiesel Chemical Reaction



The process of making biodiesel (methyl esters) is performed using methanol, heat and a catalyst to induce a transesterification reaction.

Image from: www.rise.org.au/info/Tech/waste/index.html

References

- [1] Huber, George W. et al. Synthesis of transportation fuels from biomass: chemistry, catalysts, and engineering. *Chemical Reviews*, Valencia, **2006**; 4044-4098, 6, 9
- [2] Kass, D. L. et al. Biomass for renewable energy, fuels and chemicals. *Fuels and Chemicals*; Academic Press, San Diego, CA **1998**
- [3] Sheehan, J. A. et al. A look back at the U.S. department of energy aquatic species program – biodiesel from algae. *Golden, CO 1998*
- [4] Yusuf, Christi. Biodiesel from microalgae. *Biotechnology Advances*, Palmerston North, New Zealand **2007**; 29-306, 25, 3
- [5] Denac, M. Modeling dynamic experiments on the anaerobic degradation of molasses wastewater. *Biotechnology and Bioengineering*; Zurich, Switzerland **2004**; 1-10, 31, 1